

1248665

THE UNITED STATES OF AMERICA

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UNITED STATES DEPARTMENT OF COMMERCE

United States Patent and Trademark Office

November 12, 2004

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APPLICATION NUMBER: 60/509,820

FILING DATE: *October 08, 2003*

RELATED PCT APPLICATION NUMBER: *PCT/US04/33540*

Certified by



Jon W Dudas

Acting Under Secretary of Commerce
for Intellectual Property
and Acting Director of the U.S.
Patent and Trademark Office



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PROVISIONAL APPLICATION FOR PATENT COVER SHEET

This is a request for filing a PROVISIONAL APPLICATION FOR PATENT under 37 CFR 1.53(c).

Express Mail Label No. _____

INVENTOR(S)					
Given Name (first and middle [if any])		Family Name or Surname		Residence (City and either State or Foreign Country)	
Brian E.		Boehmer		Bartlett, Tennessee	
Joshua D.		Crews		Memphis, Tennessee	
John P.		Erspamer		Lakeland, Tennessee	
Additional inventors are being named on the _____ separately numbered sheets attached hereto					
TITLE OF THE INVENTION (500 characters max)					
LATENT CARBON DIOXIDE GENERATING MATERIAL					
Direct all correspondence to: CORRESPONDENCE ADDRESS					
<input checked="" type="checkbox"/> Customer Number: 07278					
OR					
<input type="checkbox"/> Firm or Individual Name: Sandra S. Lee DARBY & DARBY P.C.					
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City: New York		State: NY		Zip: 10150-5257	
Country: US		Telephone: (212) 527-7700		Fax: (212) 753-6237	
ENCLOSED APPLICATION PARTS (check all that apply)					
<input checked="" type="checkbox"/> Specification Number of Pages: 12		<input type="checkbox"/> CD(s), Number: _____			
<input checked="" type="checkbox"/> Drawing(s) Number of Sheets: 1		<input type="checkbox"/> Other: _____			
<input checked="" type="checkbox"/> Application Data Sheet. See 37 CFR 1.76		(specify): _____			
METHOD OF PAYMENT OF FILING FEES FOR THIS PROVISIONAL APPLICATION FOR PATENT					
<input type="checkbox"/> Applicant claims small entity status. See 37 CFR 1.27.					
<input checked="" type="checkbox"/> A check or money order is enclosed to cover the filing fees.					
<input checked="" type="checkbox"/> The Director is hereby authorized to charge filing fees or credit any overpayment to Deposit Account Number: 04-0100					
<input type="checkbox"/> Payment by credit card. Form PTO-2038 is attached.					
FILING FEE AMOUNT (\$): 160.00					
The invention was made by an agency of the United States Government or under a contract with an agency of the United States Government.					
<input checked="" type="checkbox"/> No <input type="checkbox"/> Yes, the name of the U.S. Government agency and the Government contract number are: _____					

Respectfully submitted,

[Page 1 of 1]

Date: October 8, 2003

SIGNATURE
TYPED OR
PRINTED NAME
TELEPHONE

Sandra Lee

Sandra S. Lee

(212) 527-7735

REGISTRATION NO.
(if appropriate)

51,932

Docket Number:

01313/0200131-US0

USE ONLY FOR FILING A PROVISIONAL APPLICATION FOR PATENT

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14023 U.S. PTO

PTO/SB/17 (10-02)

Approved for use through 10/31/2002. OMB 0651-0032

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FEE TRANSMITTAL for FY 2003 <small>Patent fees are subject to annual revision.</small>		Complete if Known																																																																																																																																																																																															
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ADDITIONAL FEES <table border="1" style="width: 100%; border-collapse: collapse;"><thead><tr><th colspan="2">Large Entity</th><th colspan="2">Small Entity</th><th rowspan="2">Fee Description</th><th rowspan="2">Fee Paid</th></tr><tr><th>Fee Code</th><th>Fee (\$)</th><th>Fee Code</th><th>Fee (\$)</th></tr></thead><tbody><tr><td>1051</td><td>130</td><td>2051</td><td>65</td><td>Surcharge - late filing fee or oath</td><td></td></tr><tr><td>1052</td><td>50</td><td>2052</td><td>25</td><td>Surcharge - late provisional filing fee or cover sheet</td><td></td></tr><tr><td>1053</td><td>130</td><td>1053</td><td>130</td><td>Non-English specification</td><td></td></tr><tr><td>1812</td><td>2,520</td><td>1812</td><td>2,520</td><td>For filing a request for <i>ex parte</i> reexamination</td><td></td></tr><tr><td>1804</td><td>920*</td><td>1804</td><td>920*</td><td>Requesting publication of SIR prior to Examiner action</td><td></td></tr><tr><td>1805</td><td>1,840*</td><td>1805</td><td>1,840*</td><td>Requesting publication of SIR after Examiner action</td><td></td></tr><tr><td>1251</td><td>110</td><td>2251</td><td>55</td><td>Extension for reply within first month</td><td></td></tr><tr><td>1252</td><td>400</td><td>2252</td><td>200</td><td>Extension for reply within second month</td><td></td></tr><tr><td>1253</td><td>920</td><td>2253</td><td>460</td><td>Extension for reply within third month</td><td></td></tr><tr><td>1254</td><td>1,440</td><td>2254</td><td>720</td><td>Extension for reply within fourth month</td><td></td></tr><tr><td>1255</td><td>1,960</td><td>2255</td><td>980</td><td>Extension for reply within fifth month</td><td></td></tr><tr><td>1401</td><td>320</td><td>2401</td><td>160</td><td>Notice of Appeal</td><td></td></tr><tr><td>1402</td><td>320</td><td>2402</td><td>160</td><td>Filing a brief in support of an appeal</td><td></td></tr><tr><td>1403</td><td>280</td><td>2403</td><td>140</td><td>Request for oral hearing</td><td></td></tr><tr><td>1451</td><td>1,510</td><td>1451</td><td>1,510</td><td>Petition to institute a public use proceeding</td><td></td></tr><tr><td>1452</td><td>110</td><td>2452</td><td>55</td><td>Petition to revive - unavoidable</td><td></td></tr><tr><td>1453</td><td>1,280</td><td>2453</td><td>640</td><td>Petition to revive - unintentional</td><td></td></tr><tr><td>1501</td><td>1,280</td><td>2501</td><td>640</td><td>Utility issue fee (or reissue)</td><td></td></tr><tr><td>1502</td><td>460</td><td>2502</td><td>230</td><td>Design issue fee</td><td></td></tr><tr><td>1503</td><td>620</td><td>2503</td><td>310</td><td>Plant issue fee</td><td></td></tr><tr><td>1460</td><td>130</td><td>1460</td><td>130</td><td>Petitions to the Commissioner</td><td></td></tr><tr><td>1807</td><td>50</td><td>1807</td><td>50</td><td>Processing fee under 37 CFR 1.17(q)</td><td></td></tr><tr><td>1808</td><td>180</td><td>1808</td><td>180</td><td>Submission of Information Disclosure Stmt</td><td></td></tr><tr><td>8021</td><td>40</td><td>8021</td><td>40</td><td>Recording each patent assignment per property (times number of properties)</td><td></td></tr><tr><td>1809</td><td>740</td><td>2809</td><td>370</td><td>Filing a submission after final rejection (37 CFR 1.129(a))</td><td></td></tr><tr><td>1810</td><td>740</td><td>2810</td><td>370</td><td>For each additional invention to be examined (37CFR 1.129(b))</td><td></td></tr><tr><td>1801</td><td>740</td><td>2801</td><td>370</td><td>Request for Continued Examination (RCE)</td><td></td></tr><tr><td>1802</td><td>900</td><td>1802</td><td>900</td><td>Request for expedited examination of a design application</td><td></td></tr><tr><td colspan="6">Other fee (specify) _____</td></tr><tr><td colspan="4">*Reduced by Basic Filing Fee Paid</td><td colspan="2">SUBTOTAL (3) (\$) 0.00</td></tr></tbody></table>		Large Entity		Small Entity		Fee Description	Fee Paid	Fee Code	Fee (\$)	Fee Code	Fee (\$)	1051	130	2051	65	Surcharge - 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SUBMITTED BY Name (Print/Type) Sandra S. Lee Signature <i>Sandra Lee</i>		Complete (if applicable) Registration No. (Attorney/Agent) 51,932 Telephone (212) 527-7735 Date October 8, 2003																																																																																																																																																																																															

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Specification (12 pages);
Claims (7 pages, claims 1-49);
Drawings (1 sheet);
Provisional Patent Application Transmittal (1 page);
Application Data Sheet (3 pages);
Fee Transmittal for FY 2003 (1 page); and
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LATENT CARBON DIOXIDE GENERATING MATERIAL

FIELD OF THE INVENTION

This invention relates to nonwoven materials containing a chemical composition
capable of generating carbon dioxide when activated.

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BACKGROUND OF THE INVENTION

It is known in the state of the art that the freshness of fruits and vegetables and
other perishable food and agricultural materials can be preserved by the use of an
environment rich in carbon dioxide. US Patent No. 5,489,399, hereby incorporated
herein by reference in its entirety, discloses various materials useful in carbon dioxide gas
generating compositions. This disclosure teaches that the carbon dioxide gas generating
composition is usually packed into a bag made of nonwoven fabric, woven fabric, film
formed with fine pores not passing a liquid therethrough or composite sheet of such
materials and then placed in a package with fruit or vegetables. The disclosure teaches
that either the alkali carbonate or the acid of the composition must be of low solubility,
since if both are highly soluble the composition itself became liquid.

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SUMMARY OF THE INVENTION

It would be advantageous to have a delivery system for carbon dioxide gas
generating compositions without the disadvantages of the prior art systems.

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This invention provides a nonwoven carrier for carbon dioxide gas generating compositions which is a latent carbon dioxide gas generating material capable of activation under conditions of use so that a local environment rich in carbon dioxide may be established. As opposed to the prior art where all or most of the gas generating composition for a given container of produce is placed in a single location within the container, this invention provides a distribution of the particles or powder of the carbon dioxide gas generating composition throughout a substantial portion of the nonwoven carrier, which may then be installed in the container in any manner desired. Regardless of how the material is placed in a container, horizontal, vertical or otherwise, most of the powder is retained within the nonwoven carrier. The gas generating composition is activated by the absorption of water, which facilitates the chemical reaction of the acid and the base to produce carbon dioxide.

The prior art issue of the gas generating composition itself becoming liquid, either before, during or after reaction to form carbon dioxide, is obviated by distribution of the composition throughout a substantial portion of the nonwoven structure, which is produced from absorbent material which will absorb and retain any liquefied aspect of the composition within the material.

This invention provides a latent carbon dioxide gas generating material comprising:

(A) layer 1 containing fiber and a binder, having a basis weight of from about 25 gsm to about 100 gsm and having inner and outer surfaces,

(B) layer 2 containing a carbon dioxide gas generating composition, having a basis weight of from about 5 gsm to about 300 gsm and having upper and lower surfaces, and

(C) layer 3 containing fiber and a binder, having a basis weight of from about 25 gsm to about 100 gsm and having inner and outer surfaces,

where the inner surface of layer 1 is in contact with the upper surface of layer 2 and the inner surface of layer 3 is in contact with the lower surface of layer 2.

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DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENTS

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In a preferred embodiment of the material of this invention, the fiber and binder of a layer is from about 60 weight percent to about 95 weight percent fiber and from about 5 weight percent to about 40 weight percent binder, where the weight percentages are based on the total weight of fiber and binder in the layer.

5 The binder of layer 1, layer 3 and when present in layer 2 may be a synthetic binder fiber, a latex or a combination thereof. The binder composition of the layers may be identical or different. In one embodiment it is desirable that one or more layers contain a synthetic binder fiber in combination with a latex binder. In that case, the binder of the layer is a combination of from about 25 weight percent to about 75 weight
10 percent synthetic binder fiber and from about 75 weight percent to about 25 weight percent latex where the weight percentages are based on the total weight of binder in the layer and the weight percent latex includes latex solids and fluid carrier. When a binder fiber, such as, for example, bicomponent fiber is used, it is generally mixed with fiber of the layer. When a latex binder is used, it may be applied to either or both of the surfaces
15 of a layer.

 The carbon dioxide gas generating composition of layer 2 contains a weak base and a weak acid or a salt which is convertible into an acid. Generally, the weak acid has a solubility of 0.5 g/100 g of water or greater at 30°C, and has a melting point of 30°C or greater. Examples of acids are citric acid, tartaric acid, succinic acid, fumaric acid,
20 maleic acid, adipic acid, malic acid, oxalic acid, malonic acid, glutaric acid, phthalic acid, metaphosphoric acid, or wherein the salt which is convertible into an acid is an alkali metal salt of citric acid, tartaric acid, succinic acid, fumaric acid, maleic acid, adipic acid, malic acid, oxalic acid, malonic acid, glutaric acid, phthalic acid, metaphosphoric acid or a mixture thereof. Examples of salts which are convertible into acids are aluminum
25 sulfate, calcium phosphate, alum, a double salt of an alum, potassium aluminum sulfate, sodium dihydrogen phosphate, potassium citrate, sodium maleate, potassium tartrate, sodium fumarate.

 Generally, the weak base contains anionic carbonate or hydrogen carbonate, and, generally, the base contains as a cation an alkali metal, an alkaline earth metal or a
30 transition metal. Examples of bases suitable for use in the practice of this invention are lead carbonate, calcium carbonate, barium carbonate, strontium carbonate, magnesium

carbonate, beryllium carbonate, lithium carbonate, sodium carbonate, potassium carbonate, rubidium carbonate, cesium carbonate, lead hydrogen carbonate, calcium hydrogen carbonate, barium hydrogen carbonate, strontium hydrogen carbonate, magnesium hydrogen carbonate, beryllium hydrogen carbonate, lithium hydrogen carbonate, sodium hydrogen carbonate, potassium hydrogen carbonate, rubidium hydrogen carbonate, cesium hydrogen carbonate, or a mixture thereof.

In a preferred composition the acid is citric acid and base is sodium hydrogen carbonate.

It is desirable that the ratio by weight of acid to base in the carbon dioxide gas generating composition is from about 10:1 to about 1:10, preferably, that the ratio by weight of acid to base in the carbon dioxide gas generating composition is from about 5:1 to about 1:5. It is desirable that the ratio of equivalents of acid to equivalents of base in the carbon dioxide gas generating composition is from about 5:1 to about 1:5, preferably, that the ratio of equivalent of acid to equivalents of base in the carbon dioxide gas generating composition is from about 3:2 to about 2:3

In another aspect of this invention, the latent carbon dioxide gas generating material further comprises a carrier having a basis weight of from about 10 gsm to about 40 gsm which is in contact with the outer surface of layer 1 or layer 3. The carrier may be a cellulosic tissue carrier or made of synthetic material, such as, for example, a polyolefin or polyester.

Desirably, the latent carbon dioxide gas generating material of this invention has a basis weight of from about 65 gsm to about 500 gsm. One desirable embodiment has a basis weight of from about 65 gsm to about 140 gsm, while an alternative desirable embodiment has a basis weight of from about 141 gsm to about 300 gsm, and another desirable embodiment has a basis weight of from about 301 gsm to about 500 gsm.

The carbon dioxide gas generating composition may be from about 3 weight percent to about 60 weight percent of the latent carbon dioxide gas generating material. In one embodiment, the carbon dioxide gas generating composition is from about 3 weight percent to about 20 weight percent based on the total weight of the material, while in another embodiment, the carbon dioxide gas generating composition is from about 21 weight percent to about 40 weight percent based on the total weight of the material, and

in a further embodiment, the carbon dioxide gas generating composition is from about 41 weight percent to about 60 weight percent based on the total weight of the material.

5 In one embodiment of this invention, the carbon dioxide gas generating composition contains a hydrophilic material with a hygroscopicity of from about 80 weight percent to about 250 weight percent based on the weight of the hydrophilic material at 30°C. The hydrophilic material may be a saccharide or derivative thereof, polyhydric alcohol or derivative thereof, polyacrylamide or derivative thereof, polyelectrolyte or a water absorbing polymer. Examples of hydrophilic materials are erythrose, threose, arabinose, xylose, lyxose, glucose, fructose, mannose, galactose, 10 sorbitol, gluconic acid, sucrose, lactose, maltose, dextrin, amylose and hydroxymethyl cellulose.

The carbon dioxide gas generating composition may contain a hydrophobic material, such as, for example, a wax, a fatty acid or a salt or ester of a fatty acid.

15 There are a number of manufacturing processes that can be used to produce the latent carbon dioxide gas generating material of this invention. For example, one manufacturing process is an airlaid process which is used to produce an airlaid web which is typically prepared by disintegrating or defiberizing a pulp sheet or sheets typically by a hammermill to provide substantially opened fibers. The opened fibers are then air conveyed to forming heads on the airlaid web forming machine. Several 20 manufacturers make defiberized pulp sheet airlaid web forming machines including M&J Fibretech of Denmark and Dan-Web, also of Denmark. The forming heads can include rotating or agitated drums, generally in a race track configuration which serve to maintain fiber separation until the fibers are pulled by vacuum onto a foraminous condensing drum or foraminous forming conveyor. Other fibers, such as a synthetic thermoplastic fiber, or 25 superabsorbent fiber can also be introduced to the forming head through a fiber dosing system which includes a fiber opening, a dosing unit and an air conveyor. Non-fibrous materials, such as, for example a carbon dioxide gas generating composition and various additives, such as, for example, a hydrophilic material such as a superabsorbent polymer (SAP) or a hydrophobic material in powder form, particulate or granules can also be 30 added to the forming head by a dosing system.

Typically, the airlaid web is transferred from the condensing drum or forming conveyor to a calender or other densification stage to densify the web, increase its strength and control web thickness. The fibers of the web are then bonded by application of a latex spray or foam addition system, followed by drying or curing. Alternatively, or additionally, any thermoplastic fiber present in the web can be softened or partially melted by application of heat to bond the fibers of the web. The bonded web can then be calendered a second time to increase strength or emboss the web with a design or pattern. If thermoplastic fibers are present, hot calendering can be employed to impart patterned bonding to the web. Water can be added to the web if necessary to maintain specified or desired moisture content, to minimize dusting and to reduce the buildup of static electricity.

The airlaid manufacturing system has the advantage that it is a continuous process which employs a series of unit operations to produce a unitary material. Alternatively, various elements of the material can be produced as individual components and then assembled in a converting operation.

The bulk fibers of the present invention may be natural, synthetic, or a mixture thereof. In one embodiment, the fibers may be cellulose-based pulp fibers, one or more synthetic fibers, or a mixture thereof. Any cellulose fibers known in the art, including cellulose fibers of any natural origin, such as those derived from wood pulp, may be used in a cellulosic layer. Preferred cellulose fibers include, but are not limited to, digested fibers, such as kraft, prehydrolyzed kraft, soda, sulfite, chemi-thermal mechanical, and thermo-mechanical treated fibers, derived from softwood, hardwood or cotton linters. More preferred cellulose fibers include, but are not limited to, kraft digested fibers, including prehydrolyzed kraft digested fibers. Suitable for use in this invention are the cellulose fibers derived from softwoods, such as pines, firs, and spruces. Other suitable cellulose fibers include those derived from Esparto grass, bagasse, kemp, flax and other lignaceous and cellulosic fiber sources. Suitable cellulose fibers include, but are not limited to, bleached Kraft southern pine fibers sold under the trademark FOLEY FLUFFS®, Buckeye Technologies Inc., Memphis, Tennessee. Synthetic fibers include for example polyester such as PET, nylon, polyethylene or polypropylene.

Bicomponent fibers having a core and sheath are known in the art. Many varieties are used in the manufacture of nonwoven materials, particularly those produced by airlaid techniques. Various bicomponent fibers suitable for use in the present invention are disclosed in U.S. Patents 5,372,885 and 5,456,982, both of which are hereby incorporated by reference in their entirety. Examples of bicomponent fiber manufacturers include KoSa (Salisbury, NC), Trevira (Bobingen, Germany) and ES Fiber Visions (Athens, GA).

Various geometries can be used for the bicomponent fiber of this invention, including concentric, eccentric, islands-in-the-sea, and side-by-side. The relative weight percentages of the core and sheath components of the total fiber may be varied.

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Industries, West Paterson, New Jersey), which is an anionic surfactant. Other classes of emulsion polymer binders such as styrene-butadiene and acrylic binders may also be used.

Various materials, structures and manufacturing processes useful in the practice of this invention are disclosed in U.S. Patent No.s 6,241,713; 6,353,148; 6,353,148; 6,171,441; 6,159,335; 5,695,486; 6,344,109; 5,068,079; 5,269,049; 5,693,162; 5,922,163; 6,007,653; and 6,355,079; and in U.S. Patent applications with serial numbers and filing dates, 09/211,935 filed 12/15/98; 09/232,783 filed 1/19/99; 09/719,338 filed 1/17/01; 09/475,850 filed 12/30/99; 09/469,930 filed 12/21/99; 09/578,603 filed 5/25/00; 05/593,409 filed 6/14/00; 09/325,764 filed 6/8/99 allowed; 09/774,248 filed 1/30/01; and 09/854,179 filed 5/11/01, all of which are hereby incorporated by reference in their entirety.

EXAMPLES

Example 1:

Handsheets were formed on a lab airlaid forming device in two layers. The pads were built on top of a Spunbond-Meltblown-Meltblown-Spunbond (SMMS) polypropylene hydrophobic carrier having a basis weight of 17 gsm, NW 58 from First Quality Nonwovens, Great Neck, NY.

Layer 1 was a blend of 55 gsm cellulose fluff, FOLEY FLUFFS®, Buckeye, Perry, FL, and 9.7 gsm 2.0 dpf x 6mm bicomponent fiber with a polyester core and a polyethylene sheath, F55 from KoSa, Charlotte, NC, which was deposited directly on the carrier. Then 5.6 gsm of a pre-mixed powder of sodium bicarbonate and acetylsalicylic acid from CO₂ Technologies, West Des Moines, IA, was added by hand. The powder had an 88:12 base:acid powder content ratio by weight. The carrier side was sprayed with 11 gsm of an ethylene vinyl acetate latex with 20 percent solids, EP-1188 from Air Products, Allentown PA.

Layer 2 was a blend of 110 gsm cellulose fluff, FOLEY FLUFFS®, and 19.4 gsm bicomponent fiber, KoSa F55, 2.0 dpf x 6mm, which was deposited on top of the powder. The materials were pressed with shims at 5.08 mm (200 mils) thickness and 239.2 kPa

(5000 psi) pressure with platens heated to 150°C for 1 minute. The pads were then cured for 15 minutes at 145°C.

Example 2:

- 5 Handsheets were formed as described in Example 1, except that 11.3 gsm of a pre-mixed powder of sodium bicarbonate and acetylsalicylic acid with an 88:12 base:acid powder content ratio by weight from CO₂ Technologies, was added by hand.

Example 3:

- 10 Handsheets were formed as described in Example 1, except that 22.5 gsm of a pre-mixed powder of sodium bicarbonate and acetylsalicylic acid with an 88:12 base:acid powder content ratio by weight from CO₂ Technologies, was added by hand.

Example 4:

- 15 Handsheets were formed as described in Example 1, except that 33.8 gsm of a pre-mixed powder of sodium bicarbonate and acetylsalicylic acid with an 88:12 base:acid powder content ratio by weight from CO₂ Technologies, was added by hand.

- 20 Examples 1-4 were submitted for a gas generation study. A graph of the outcome of that study appears as Figure 1.

Example 5:

- 25 Nonwoven material was made on a Dan-Web type pilot airlaid machine using three forming heads to deposit three layers on a Spunbond-Meltblown-Meltblown-Spunbond (SMMS) polypropylene hydrophobic carrier having a basis weight of 17 gsm, NW 58 from First Quality Nonwovens, Great Neck, NY.

- 30 Layer 1 was a blend of 29.2 gsm cellulose fluff, FOLEY FLUFFS® and 5.1 gsm of a 1.7 dtex x 4 mm bicomponent fiber with a polypropylene core and a polyethylene sheath, AL-Adhesion-C, Fiber Visions Covington, GA, was deposited on the carrier. This layer was sprayed with 5 gsm of Airflex 192, an ethylene vinyl acetate latex used at 15 percent solids, from Air Products, Allentown, PA.

Layer 2 was a blend of 29.2 gsm of FOLEY FLUFFS®, 5.1 gsm Fiber Visions bicomponent fiber, AL-Adhesion-C, 1.7 dtex x 4 mm, and 22.5 gsm of a blend of sodium bicarbonate from Church&Dwight, Princeton, NJ, and citric acid powders. The blend ratio of the powder was 88:12 sodium bicarbonate: citric acid by weight.

5 Layer 3 was a blend of 29.2 gsm FOLEY FLUFFS® and 5.1 gsm Fiber Visions bicomponent fiber, AL-Adhesion-C, 1.7 dtex x 4 mm. This layer was sprayed with 2.5 gsm of Airflex 192, 15% solids. The material was then pressed to a target caliper of 2.74 mm.

10 **Example 6:**

Nonwoven material was made as in Example 5, except that in Layer 1 the fiber blend contained 36.3 gsm of FOLEY FLUFFS® and 6.3 gsm of Fiber Visions bicomponent fiber, AL-Adhesion-C, Layer 2 used 36.3 gsm of FOLEY FLUFFS® and 6.3 gsm of Fiber Visions AL-Adhesion-C, and 22.5 gsm of the bicarbonate citric acid blend and Layer 3 used 36.3 gsm of FOLEY FLUFFS® and 6.3 gsm of Fiber Visions bicomponent fiber, AL-Adhesion-C. Samples were pressed to a target caliper of 3.16 mm.

Example 7:

20 Nonwoven material was made as in Example 5, except that in Layer 1, the fiber blend contained 43.4 gsm of FOLEY FLUFFS® and 7.6 gsm of Fiber Visions bicomponent fiber, AL-Adhesion-C, Layer 2 used 43.6 gsm of FOLEY FLUFFS® and 7.6 gsm of Fiber Visions AL-Adhesion-C, and 22.5 gsm of the bicarbonate citric acid blend and Layer 3 used 43.4 gsm of FOLEY FLUFFS® and 7.6 gsm of Fiber Visions bicomponent fiber, AL-Adhesion-C. Samples were pressed to a target caliper of 3.61 mm.

Example 8:

Nonwoven material was made as in Example 5, except that in Layer 1, the fiber blend contained 23.4 gsm of FOLEY FLUFFS® and 5.1 gsm of a 2.2 dtex x 3 mm bicomponent fiber with a polyethyleneterephthalate core and a polyethylene sheath, Merge 1663 from Trevira, Frankfurt am Main, Germany, Layer 2 used 43.6 gsm of FOLEY FLUFFS® and 5.1 gsm Merge 1663, and 22.5 gsm of a bicarbonate citric acid blend with base to acid ratio 4:1 by weight and Layer 3 used 43.4 gsm of FOLEY FLUFFS® and 5.1 gsm Merge 1663. Samples were pressed to a target caliper of 2.74 mm. A sample of the material was activated with water and carbon dioxide gas was generated.

Example 9:

Nonwoven material was made as in Example 8, except that in Layer 1 the fiber blend contained 26.7 gsm of FOLEY FLUFFS® and 7.8 gsm of Merge 1663, Layer 2 used 26.7 gsm of FOLEY FLUFFS® and 7.8 gsm Merge 1663, and 102 gsm of the bicarbonate citric acid blend and Layer 3 used 26.7 gsm of FOLEY FLUFFS® and 7.8 gsm Merge 1663. Samples were pressed to a target caliper of 2.8 mm. A sample of the material was activated with water and carbon dioxide gas was generated.

Example 10:

Nonwoven material was made as in Example 8, except that in Layer 1, the fiber blend contained 40 gsm of FOLEY FLUFFS® and 9.4 gsm of Merge 1663, Layer 2 used 40 gsm of FOLEY FLUFFS® and 9.4 gsm Merge 1663, and 102 gsm of the bicarbonate citric acid blend and Layer 3 used 40 gsm of FOLEY FLUFFS® and 9.4 gsm Merge 1663. Samples were pressed to a target caliper of 3.3 mm. A sample of the material was activated with water and carbon dioxide gas was generated.

CLAIMS

What is claimed is:

1. A latent carbon dioxide gas generating material comprising:
 - (A) layer 1 containing fiber and a binder, having a basis weight of from about 25 gsm to about 100 gsm and having inner and outer surfaces,
 - (B) layer 2 containing a carbon dioxide gas generating composition, having a basis weight of from about 5 gsm to about 300 gsm and having upper and lower surfaces, and
 - (C) layer 3 containing fiber and a binder, having a basis weight of from about 25 gsm to about 100 gsm and having inner and outer surfaces,
 where the inner surface of layer 1 is in contact with the upper surface of layer 2 and the inner surface of layer 3 is in contact with the lower surface of layer 2.
2. The material of claim 1, further comprising a carrier having a basis weight of from about 10 gsm to about 40 gsm which is in contact with the outer surface of layer 1.
3. The material of one of the previous claims, wherein the fiber of layer 1 is cellulosic fiber, synthetic fiber or a mixture thereof.
4. The material of claim 3, wherein the fiber of layer 1 is cellulosic fiber.
5. The material of one of the previous claims, wherein the binder of layer 1 is a synthetic binder fiber, a latex or a combination thereof.
6. The material of claim 5, wherein the synthetic binder fiber of layer 1 is a bicomponent fiber.
7. The material of one of the previous claims, wherein layer 1 contains from about 60 weight percent to about 95 weight percent fiber and from about 5 weight percent

to about 40 weight percent binder, where the weight percentages are based on the total weight of fiber and binder in the layer.

8. The material of one of the previous claims, wherein layer 1 contains a synthetic binder fiber in combination with a latex binder.

9. The material of claim 8, wherein the binder of layer 1 is a combination of from about 25 weight percent to about 75 weight percent synthetic binder fiber and from about 75 weight percent to about 25 weight percent latex where the weight percentages are based on the total weight of binder and the weight percent latex includes latex solids and fluid carrier.

10. The material of one of the previous claims, wherein layer 2 contains a carbon dioxide gas generating composition comprising a weak base and a weak acid.

11. The material of claim 10, wherein the weak acid has a solubility of 0.5 g/100 g of water or greater at 30°C, and has a melting point of 30°C or greater.

12. The material of one of the previous claims, wherein layer 2 contains a salt which is convertible into an acid.

13. The material of claim 11, wherein the acid is citric acid, tartaric acid, succinic acid, fumaric acid, maleic acid, adipic acid, malic acid, oxalic acid, malonic acid, glutaric acid, phthalic acid, metaphosphoric acid, or wherein the salt which is convertible into an acid is an alkali metal salt of citric acid, tartaric acid, succinic acid, fumaric acid, maleic acid, adipic acid, malic acid, oxalic acid, malonic acid, glutaric acid, phthalic acid, metaphosphoric acid or a mixture thereof.

14. The material of claim 10, wherein the weak base contains anionic carbonate or hydrogen carbonate.

15. The material of claim 14, wherein the base contains as a cation an alkali metal, an alkaline earth metal or a transition metal.

16. The material of claim 15, wherein the base is, lead carbonate, calcium carbonate, barium carbonate, strontium carbonate, magnesium carbonate, beryllium carbonate, lithium carbonate, sodium carbonate, potassium carbonate, rubidium carbonate, cesium carbonate, lead hydrogen carbonate, calcium hydrogen carbonate, barium hydrogen carbonate, strontium hydrogen carbonate, magnesium hydrogen carbonate, beryllium hydrogen carbonate, lithium hydrogen carbonate, sodium hydrogen carbonate, potassium hydrogen carbonate, rubidium hydrogen carbonate, cesium hydrogen carbonate, or a mixture thereof.

17. The material of claim 13, wherein the salt which is convertible into an acid is aluminum sulfate, calcium phosphate, alum, a double salt of an alum, potassium aluminum sulfate, sodium dihydrogen phosphate, potassium citrate, sodium maleate, potassium tartrate, sodium fumarate, .

18. The material of claim 10, wherein the acid is citric acid and base is sodium hydrogen carbonate.

19. The material of one of the previous claims, wherein layer 2 contains a carbon dioxide gas generating composition mixed with fiber and binder.

20. The material of claim 19, wherein the fiber of layer 2 is cellulosic fiber, synthetic fiber or a mixture thereof.

21. The material of claim 20, wherein the fiber of layer 2 is cellulosic fiber.

22. The material of one of claims 19-21, wherein the binder of layer 2 is a synthetic binder fiber, a latex or a combination thereof.

23. The material of claim 22, wherein the synthetic binder fiber of layer 2 is a bicomponent fiber.

24. The material of one of the previous claims, wherein layer 2 contains fiber and binder which are from about 60 weight percent to about 95 weight percent fiber and from about 5 weight percent to about 40 weight percent binder, where the weight percentages are based on the total weight of the fiber and binder in the layer.

24. The material of one of claims 19-23, wherein layer 2 contains a synthetic binder fiber in combination with a latex binder.

26. The material of claim 24, wherein the binder of layer 2 is a combination of from about 25 weight percent to about 75 weight percent synthetic binder fiber and from about 75 weight percent to about 25 weight percent latex where the weight percentages are based on the total weight of binder and the weight percent latex includes latex solids and fluid carrier.

27. The material of one of the previous claims, wherein the fiber of layer 3 is cellulosic fiber, synthetic fiber or a mixture thereof.

28. The material of claim 27, wherein the fiber of layer 3 is cellulosic fiber.

29. The material of one of the previous claims, wherein the binder of layer 3 is a synthetic binder fiber, a latex or a combination thereof.

30. The material of claim 29, wherein the synthetic binder fiber of layer 3 is a bicomponent fiber.

31. The material of claim 27, wherein layer 3 contains from about 60 weight percent to about 95 weight percent fiber and from about 5 weight percent to about 40

weight percent binder, where the weight percentages are based on the total weight of the layer.

32. The material of one of the previous claims, wherein layer 3 contains a synthetic binder fiber in combination with a latex binder.

33. The material of claim 32, wherein the binder of layer 3 is a combination of from about 25 weight percent to about 75 weight percent synthetic binder fiber and from about 75 weight percent to about 25 weight percent latex where the weight percentages are based on the total weight of binder and the weight percent latex includes latex solids and fluid carrier.

34. The material of one of the previous claims, wherein the basis weight of the material is from about 65 gsm to about 500 gsm.

35. The material of claim 34, wherein the basis weight of the material is from about 65 gsm to about 140 gsm.

36. The material of claim 34, wherein the basis weight of the material is from about 141 gsm to about 300 gsm.

37. The material of claim 34, wherein the basis weight of the material is from about 301 gsm to about 500 gsm.

38. The material of one of the previous claims, wherein the carbon dioxide gas generating composition is from about 3 weight percent to about 20 weight percent based on the total weight of the material.

39. The material of one of claims 1-37, wherein the carbon dioxide gas generating composition is from about 21 weight percent to about 40 weight percent based on the total weight of the material.

40. The material of one of claims 1-37, wherein the carbon dioxide gas generating composition is from about 41 weight percent to about 60 weight percent based on the total weight of the material.

41. The material of one of the previous claims, wherein the ratio by weight of acid to base in the carbon dioxide gas generating composition is from about 10:1 to about 1:10.

42. The material of claim 41, wherein the ratio by weight of acid to base in the carbon dioxide gas generating composition is from about 5:1 to about 1:5.

43. The material of one of the previous claims, wherein the ratio of equivalents of acid to equivalents of base in the carbon dioxide gas generating composition is from about 5:1 to about 1:5.

44. The material of claim 43, wherein the ratio of equivalent of acid to equivalents of base in the carbon dioxide gas generating composition is from about 3:2 to about 2:3.

45. The material of one of the previous claims, wherein the carbon dioxide gas generating composition contains a hydrophilic material with a hygroscopicity of from about 80 weight percent to about 250 weight percent based on the weight of the hydrophilic material at 30°C.

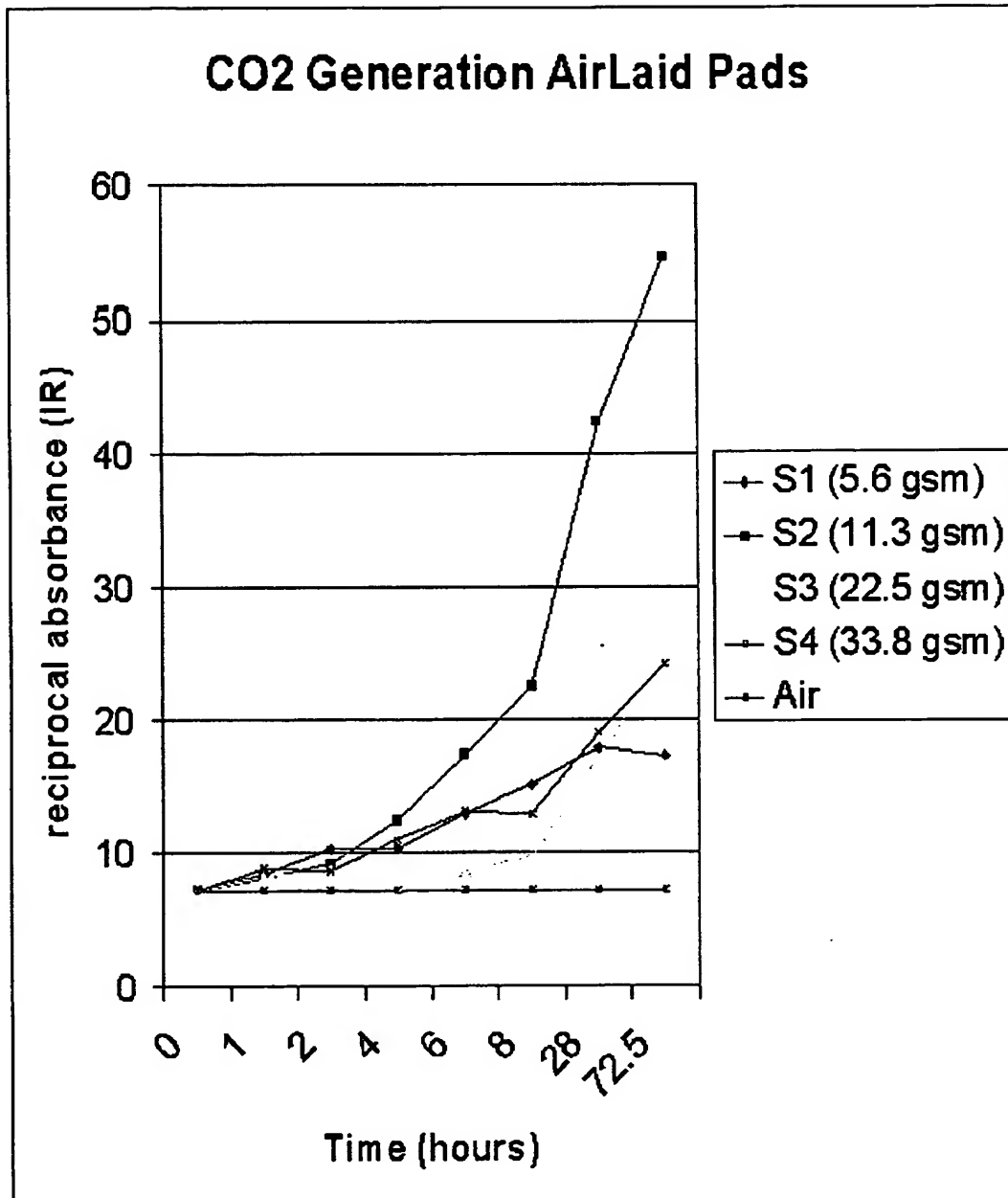
46. The material of claim 45, wherein the hydrophilic material is a saccharide or derivative thereof, polyhydric alcohol or derivative thereof, polyacrylamide or derivative thereof, polyelectrolyte or a water absorbing polymer.

47. The material of claim 46, wherein the hydrophilic material is erythrose, threose, arabinose, xylose, lyxose, glucose, fructose, mannose, galactose, sorbitol, gluconic acid, sucrose, lactose, maltose, dextrin, amylose or hydroxymethyl cellulose.

48. The material of one of the previous claims, wherein the carbon dioxide gas generating composition contains a hydrophobic material.

49. The material of claim 48, wherein the hydrophobic material is a wax, a fatty acid or a salt or ester of a fatty acid.

Figure 1



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Application Data Sheet

Application Information

Application Type::	Provisional
Subject Matter::	Utility
Suggested Group Art Unit::	N/A
CD-ROM or CD-R?::	None
Sequence submission?::	None
Computer Readable Form (CRF)?::	No
Title::	LATENT CARBON DIOXIDE GENERATING MATERIAL
Attorney Docket Number::	01313/0200131-US0
Request for Early Publication?::	No
Request for Non-Publication?::	No
Small Entity?::	No
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Secrecy Order in Parent Appl.?::	No

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